



# Molecular mobility in model heterogeneous cross-linked epoxy networks

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## Epoxy networks are an important class of thermoset polymers

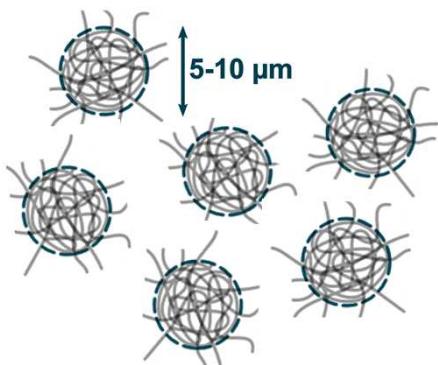
- Drawback for their use as structural materials: **Brittleness**
- **Network structure** is believed to play a crucial role in the **ultimate mechanical properties** of polymers, such as **fracture toughness**

A clear understanding of how **network structure** is correlated with thermosets' **ultimate mechanical properties** is yet to be achieved to enable the **design of polymer materials** with functionalities tailored to specific industrial applications

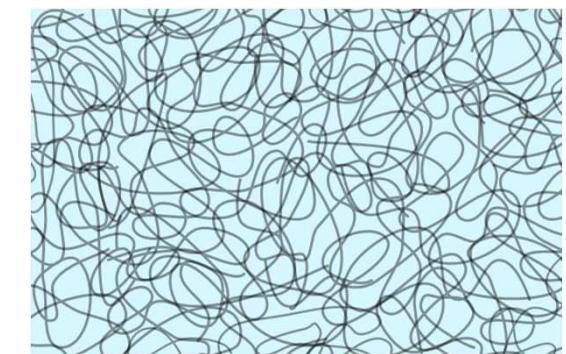
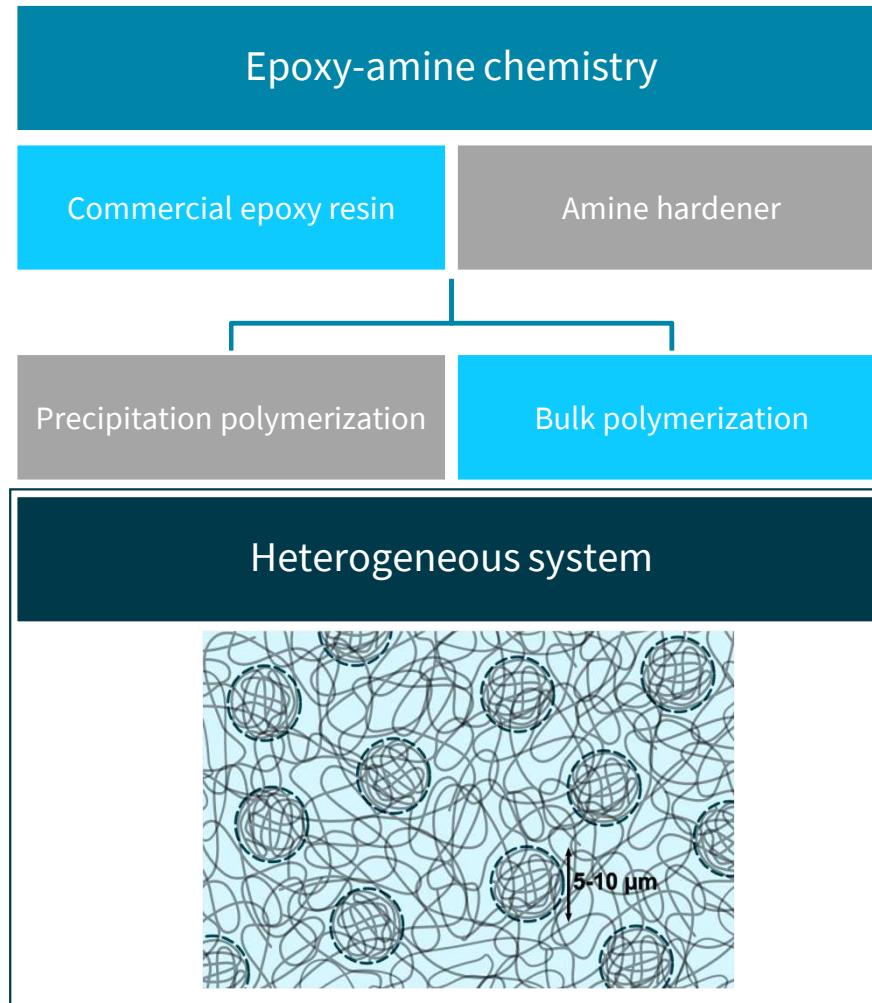
### Objective of the project:

- **To analyze the molecular mobility in polymer systems presenting an increasing degree of heterogeneity.**

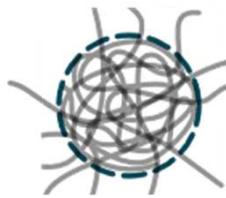
## Strategy



Crosslinked Epoxy Microgels (**CEMs**)



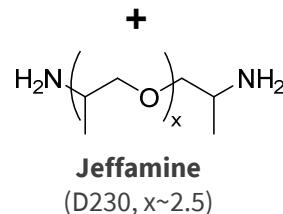
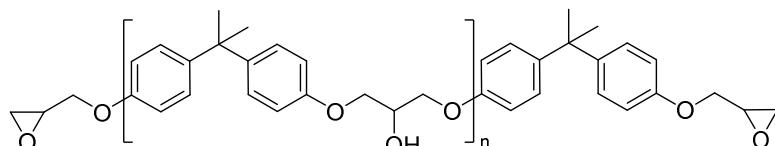
Epoxy matrix



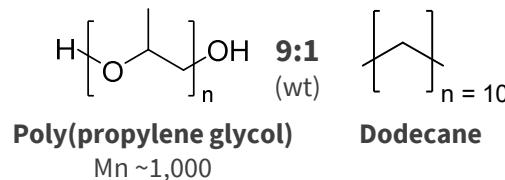
# CEMs synthesis

# The Crosslinked Epoxy Microgels (CEMs) were synthesized by precipitation polymerization<sup>1</sup>

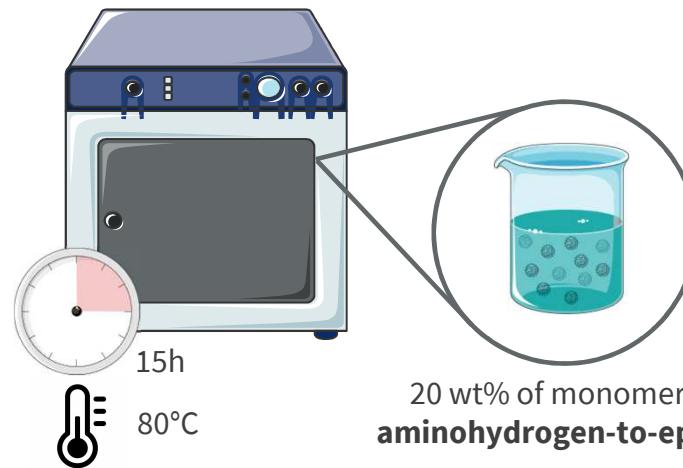
- Monomers:



- Solvent:

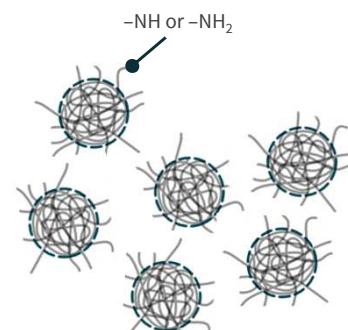


- Precipitation polymerization protocol:



20 wt% of monomers in solution  
aminohydrogen-to-epoxy ratio: 1.5

Washing, filtration,  
drying, and grinding



- Characterization:

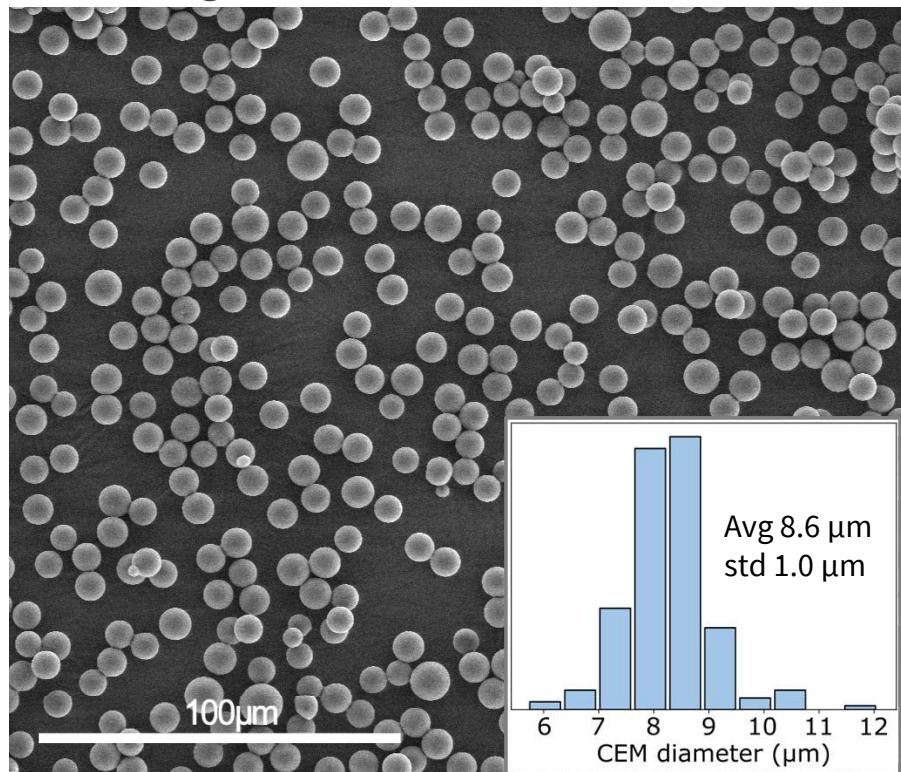
- DSC
- TGA
- IR
- SEM
- Helium pycnometry

Optimized to obtain **narrow size distribution**

1. M.L. Michon, INSA Lyon PhD, 2014.

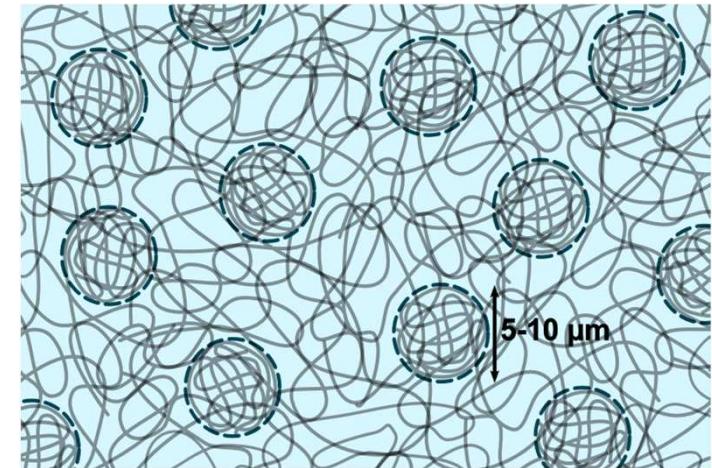
## CEMs obtained have narrow size distribution

### SEM micrographs of model CEMs:



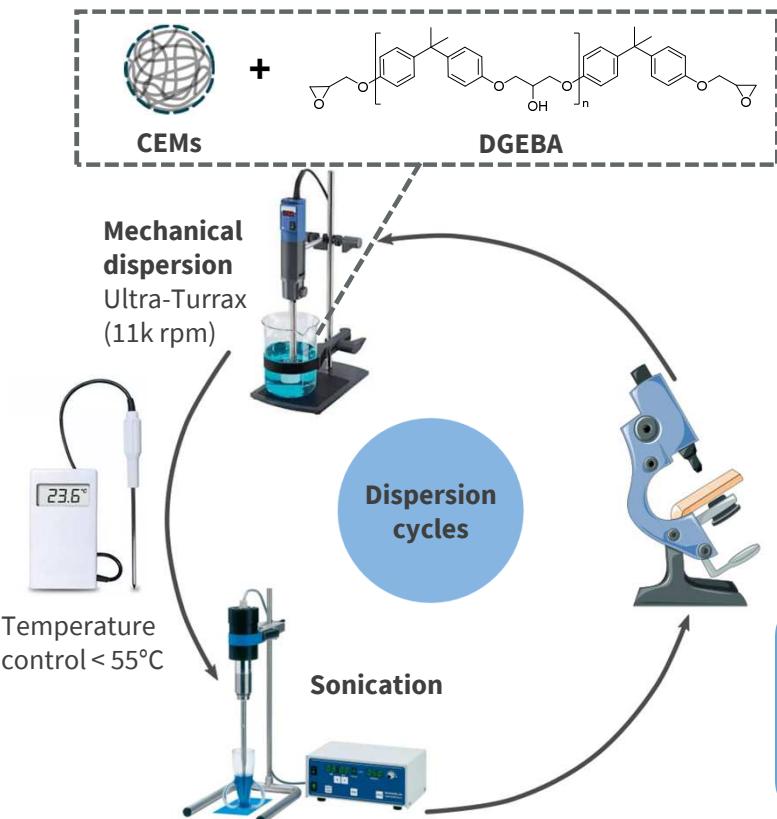
- **TGA** Solvent removal during washing protocol was confirmed
- **IR** The presence of remaining primary and secondary amine functional groups was confirmed
- **Pycnometry** Average density of CEMs :  $1.18 \text{ g/cm}^3$
- **DSC**  $T_g$  1<sup>st</sup> run:  $67^\circ\text{C}$   
 $T_g$  2<sup>nd</sup> run:  $70^\circ\text{C}$

# Obtention of the heterogeneous system

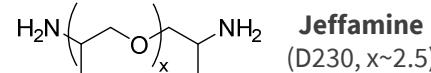


# The CEMs were dispersed in DGEBA and cured using D230 as hardener

## 1 Masterbatch (MB) preparation:



## 2 MB dilution in DGEBA followed by hardener addition



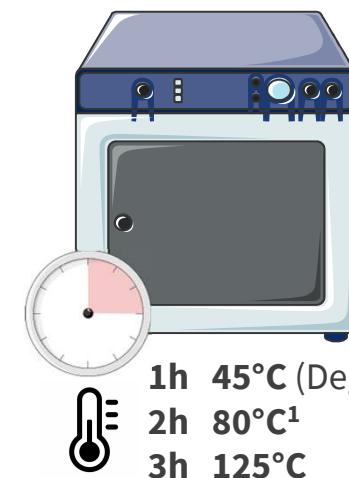
Homogenization using  
Vacuum speedmixer

30min  
800 – 2000 rpm  
30 mbar



Dispersion control  
with optical  
microscopy

## 3 Curing in vacuum oven

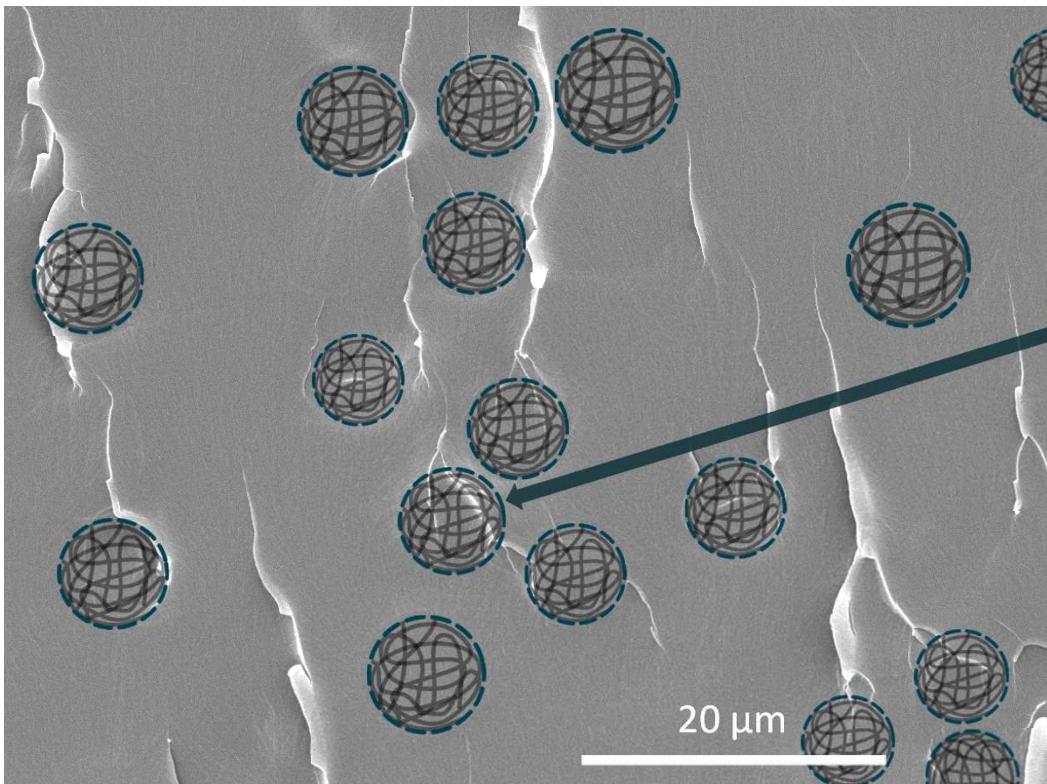


1h 45°C (Degassing step)  
2h 80°C<sup>1</sup>  
3h 125°C  
1h 200°C

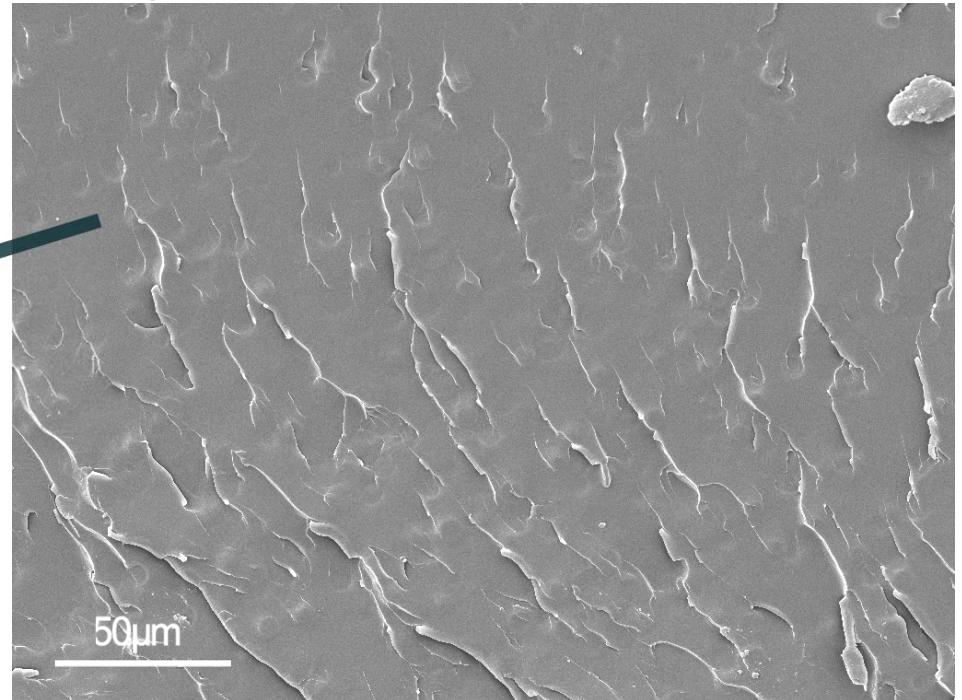
The networks were obtained as **thin films ( $\sim 500\mu\text{m}$ )** with **homogeneous thickness**; the **reference matrix film** (DGEBA/Diamine,  $a/e = 1$ ) was synthesized using the same curing protocol.

1. Nguyen, T. K. L., Livi, S., Soares, B. G., Pruvost, S., Duchet-Rumeau, J., & Gérard, J. F. (2016). Ionic liquids: A new route for the design of epoxy networks. ACS Sustainable Chemistry & Engineering

## SEM micrographs of cryogenic fracture surface of fully cured networks

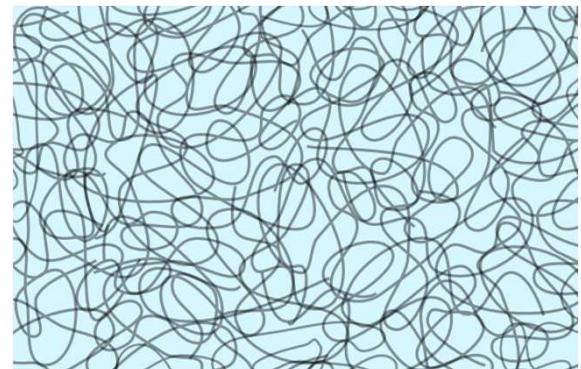
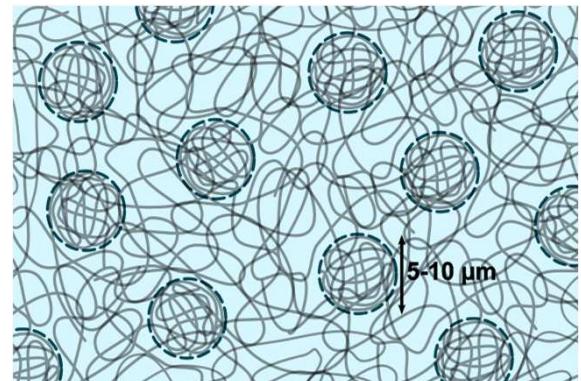


Heterogeneous network (20wt% of CEMs)

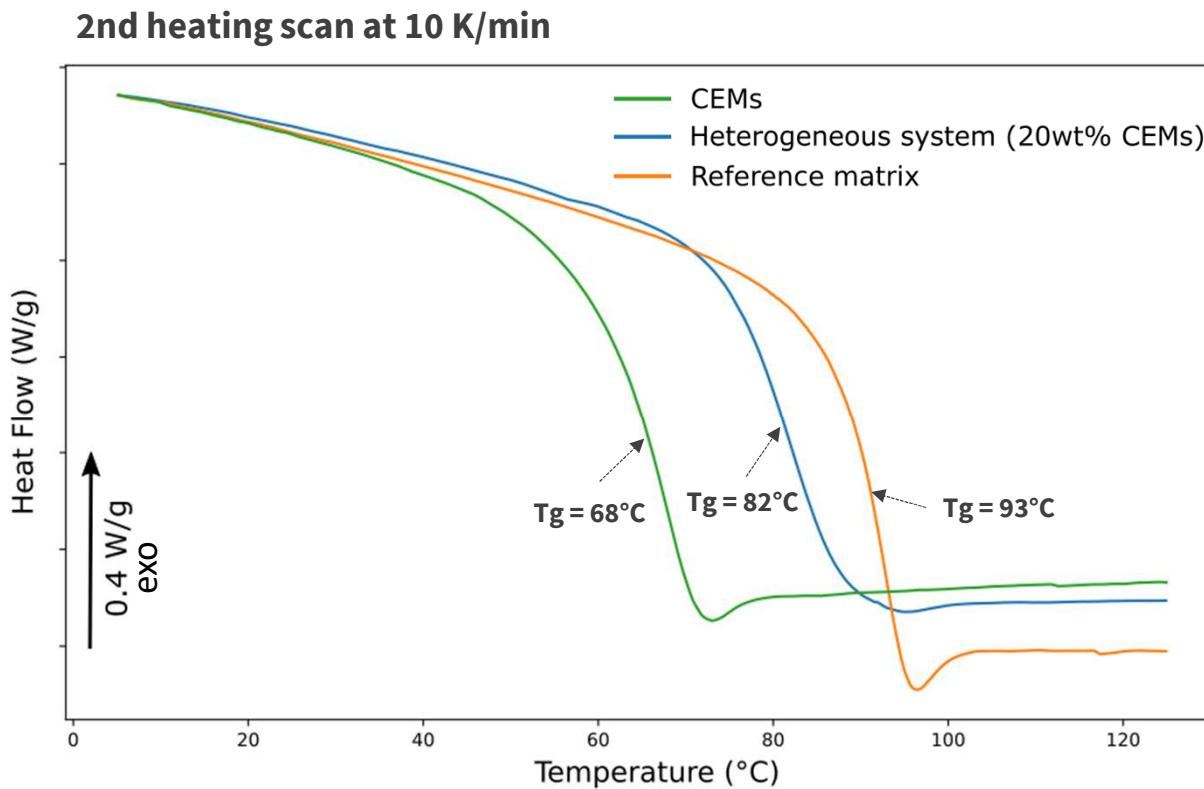


- Strong adhesion suggests that **CEMs** could be **covalently bonded** to the matrix.

# **Analysis of molecular mobility of the heterogeneous system compared to reference matrix**



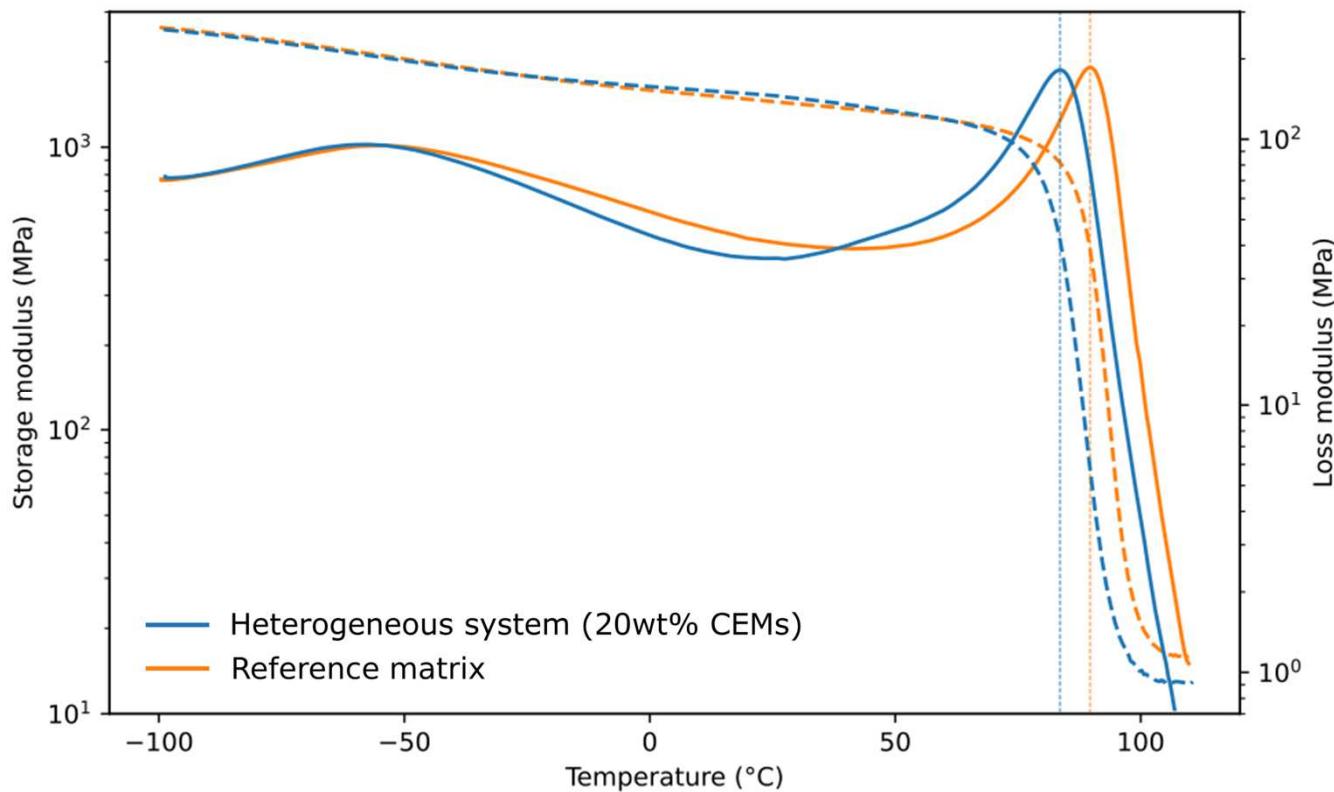
## DSC of the heterogeneous system compared to reference film and CEMs



- Only one single glass transition temperature is observed for the heterogeneous system, behaving like a homogeneous system despite the heterogeneity in SEM analysis

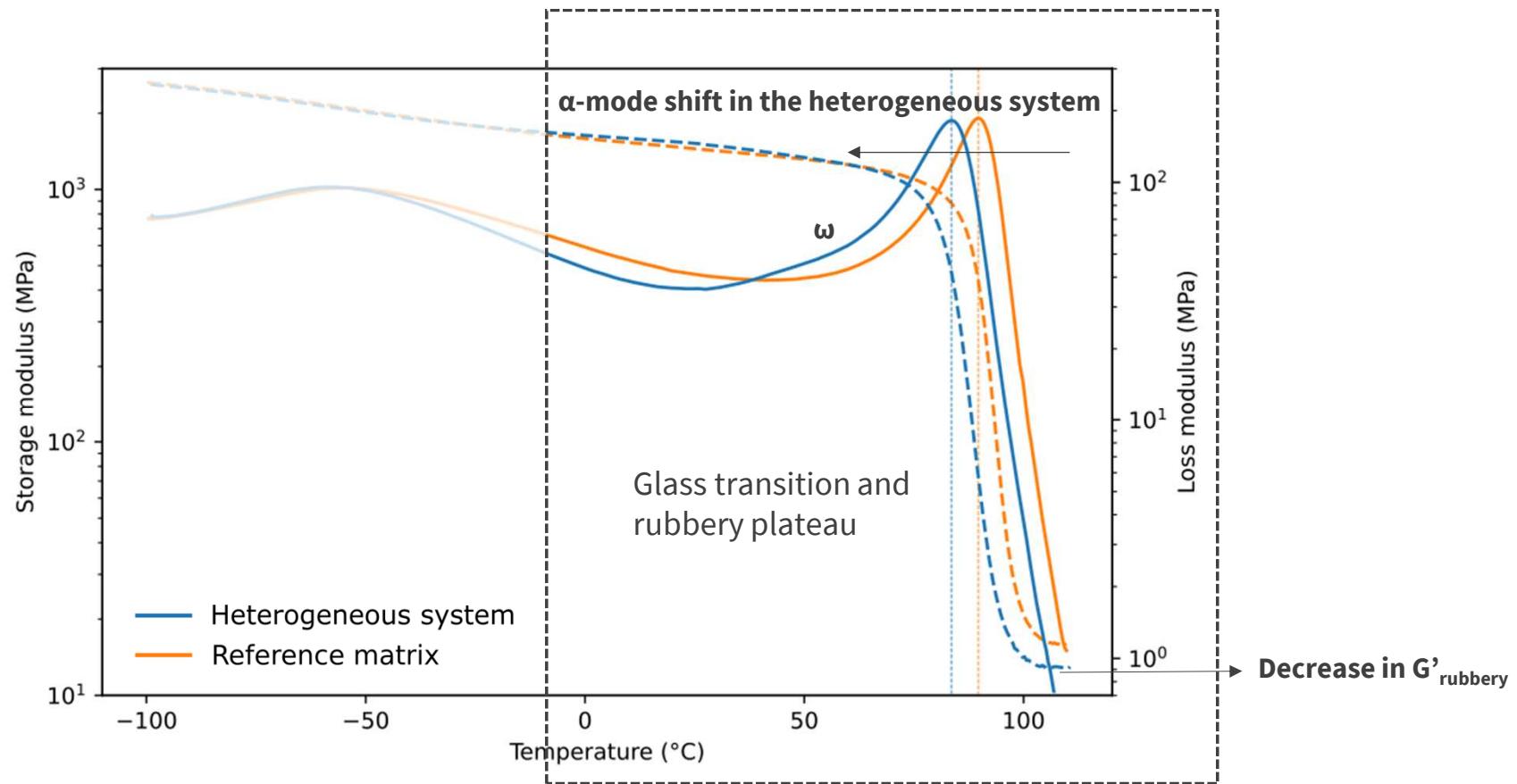
## DMA characterization of the heterogeneous system and reference matrix

- DMA storage and loss modulus curves obtained by **rectangular torsion mode at 1 Hz, 3 K/min and 0.05% strain** (2nd heating ramp)



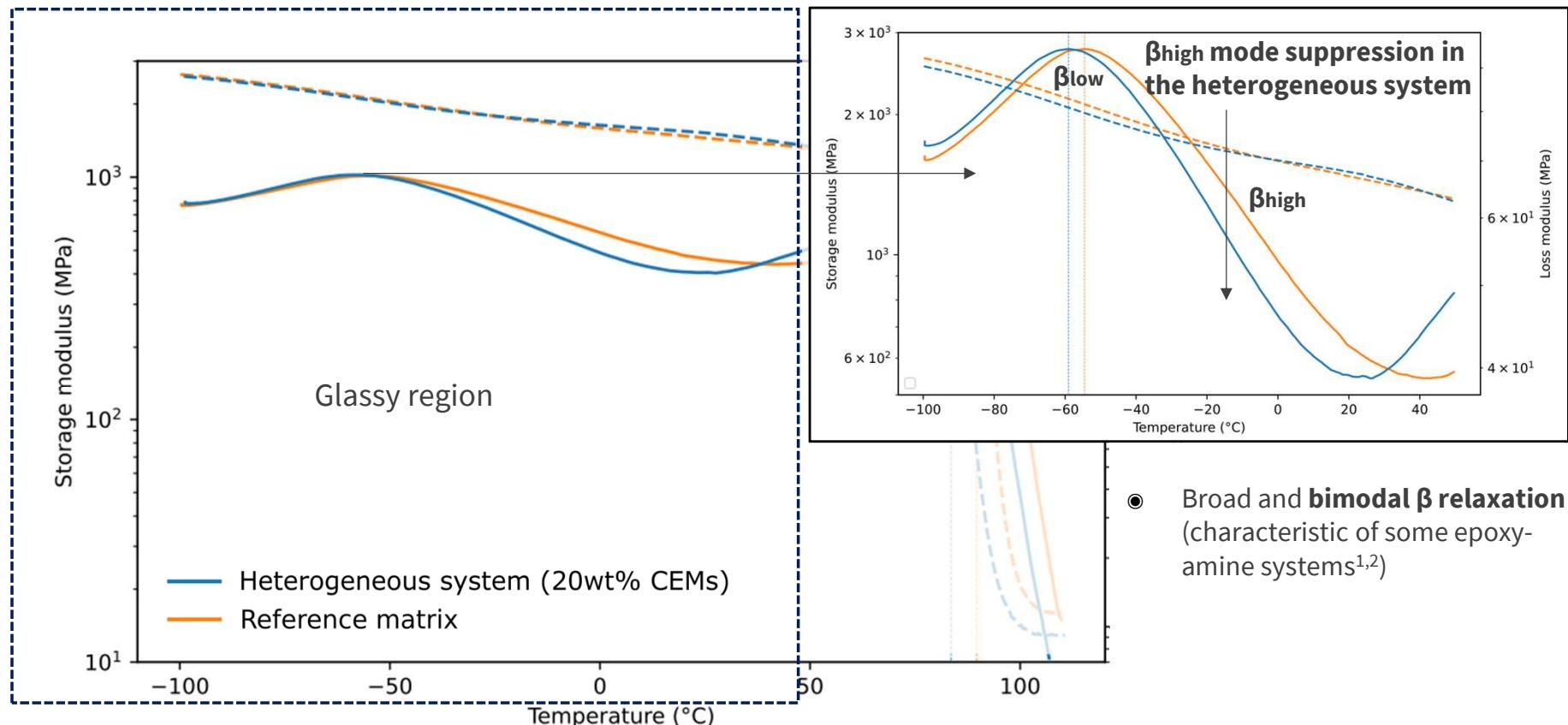
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1. Won, Y. G. et all. (1990). Internal antiplasticization in copolymer and terpolymer networks based on diepoxides, diamines and monoamines. Polymer, 31(9), 1787-1792.

2. Ramsdale-Capper, R. et all. (2018). Internal antiplasticisation in highly crosslinked amine cured multifunctional epoxy resins. Polymer, 146, 321-330.

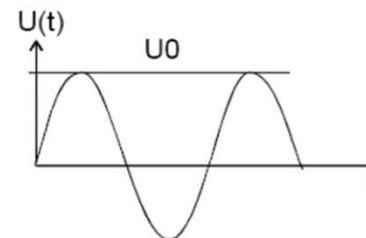
## Broadband dielectric spectroscopy (BDS) measurements



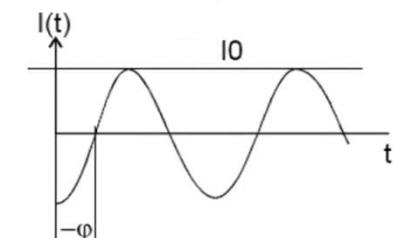
**Analysis parameters:**  
T range: -150 to 150 °C (5°C step)  
f range:  $10^6$  to  $4 \times 10^{-2}$  Hz

- Upper electrode
- Spring
- Gold plated sample
- Lower electrode

● AC voltage applied



● Measured current



Complex impedance

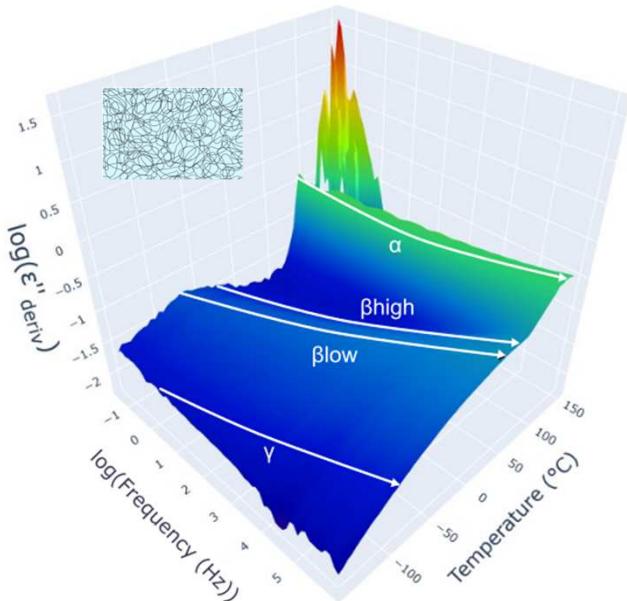
$$Z^*(\omega) = \frac{U^*(\omega)}{I^*(\omega)}$$

Complex permittivity

$$\varepsilon^*(\omega) = \varepsilon'(\omega) - i\varepsilon''(\omega) = \frac{1}{i\omega C_0 Z^*}$$

Where:  $C_0 = \frac{\varepsilon_0 A}{l}$  → Gold electrode surface  
→ Sample thickness

## Dielectric relaxation map of the reference sample

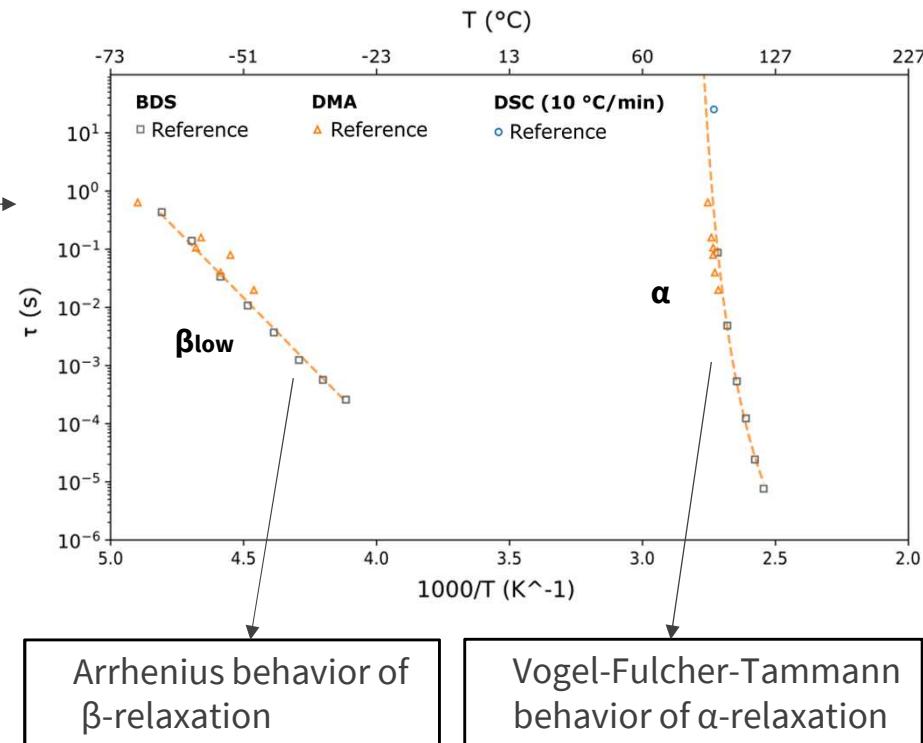


Fitted using Havriliak-Negami equation

$$\epsilon^* = \epsilon'_\infty + \frac{\epsilon'_s - \epsilon'_\infty}{(1 + (i\omega\tau_{H-N})^{\alpha_{H-N}})^{\beta_{H-N}}}$$

- DSC equivalent frequency was determined using<sup>1</sup>:

$$f_{eq} = \frac{q}{2\pi a \delta T}$$



Arrhenius behavior of  
 $\beta$ -relaxation

$$\tau^{Arr}(T) = \tau_0 e^{\frac{E_a}{RT}}$$

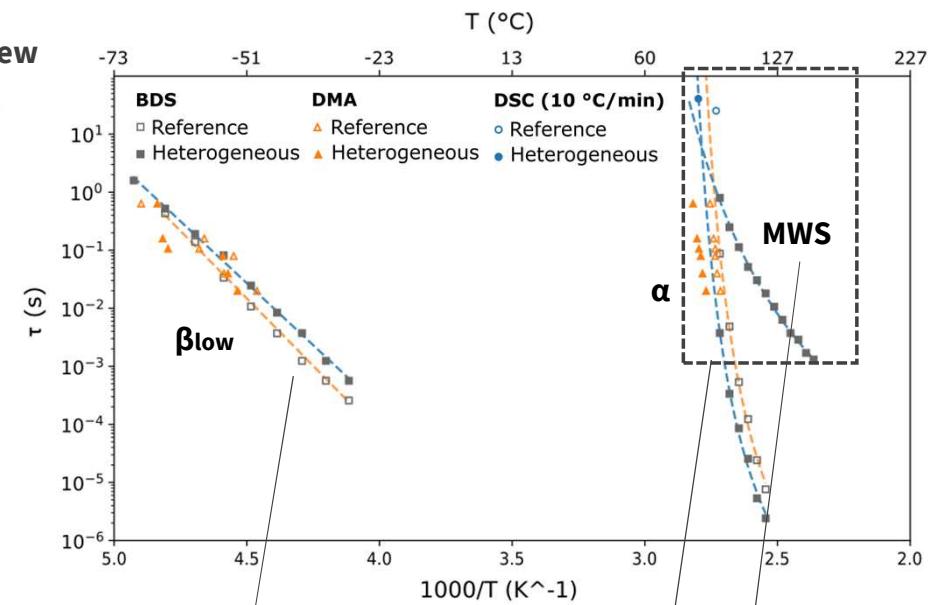
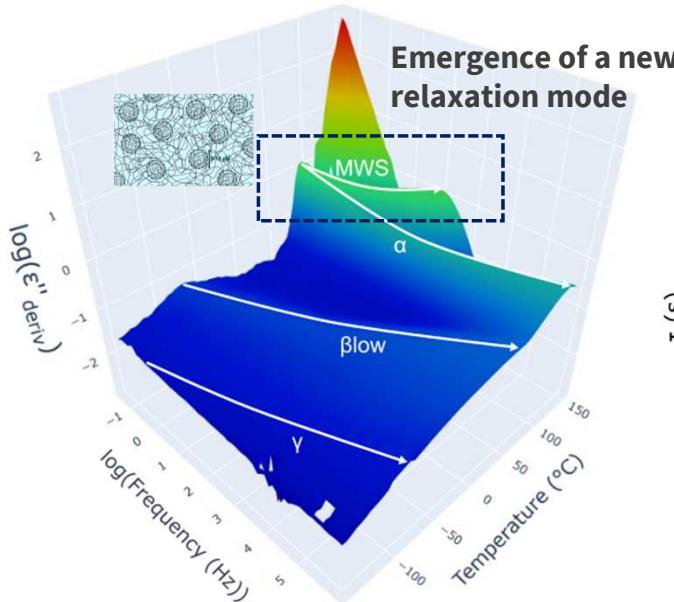
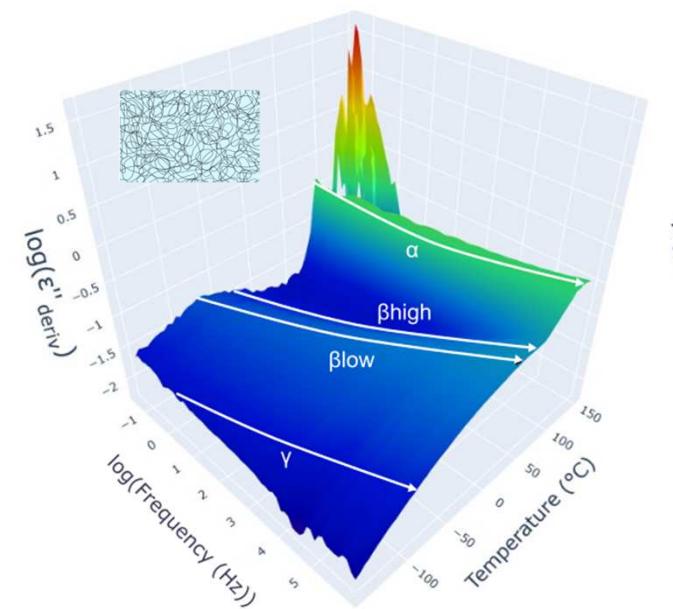
Vogel-Fulcher-Tamman  
behavior of  $\alpha$ -relaxation

$$\tau^{VFT}(T) = \tau_0 e^{\frac{1}{\alpha_f(T - T_\infty)}}$$

- Good agreement between DSC, DMA, and BDS data.

1. Donth, E.-J. Relaxation and Thermodynamics in Polymers: Glass Transition; Akademie Verlag: Berlin, 1994.

## Dielectric relaxation map of the heterogeneous network sample



- The additional relaxation mode appears only in BDS measurement, thus **no mechanical response** of the responsible entities;
- Could be ascribed to macrodipoles formed by **interfacial polarization** or Maxwell-Wagner-Sillars (**MWS**)

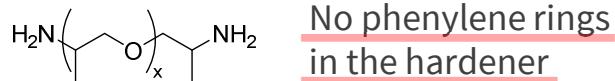
Arrhenius behavior of  
 $\beta$ -relaxation

Vogel-Fulcher-Tamman  
behavior of  $\alpha$ -relaxation

**Result of conductivity heterogeneity in the network**

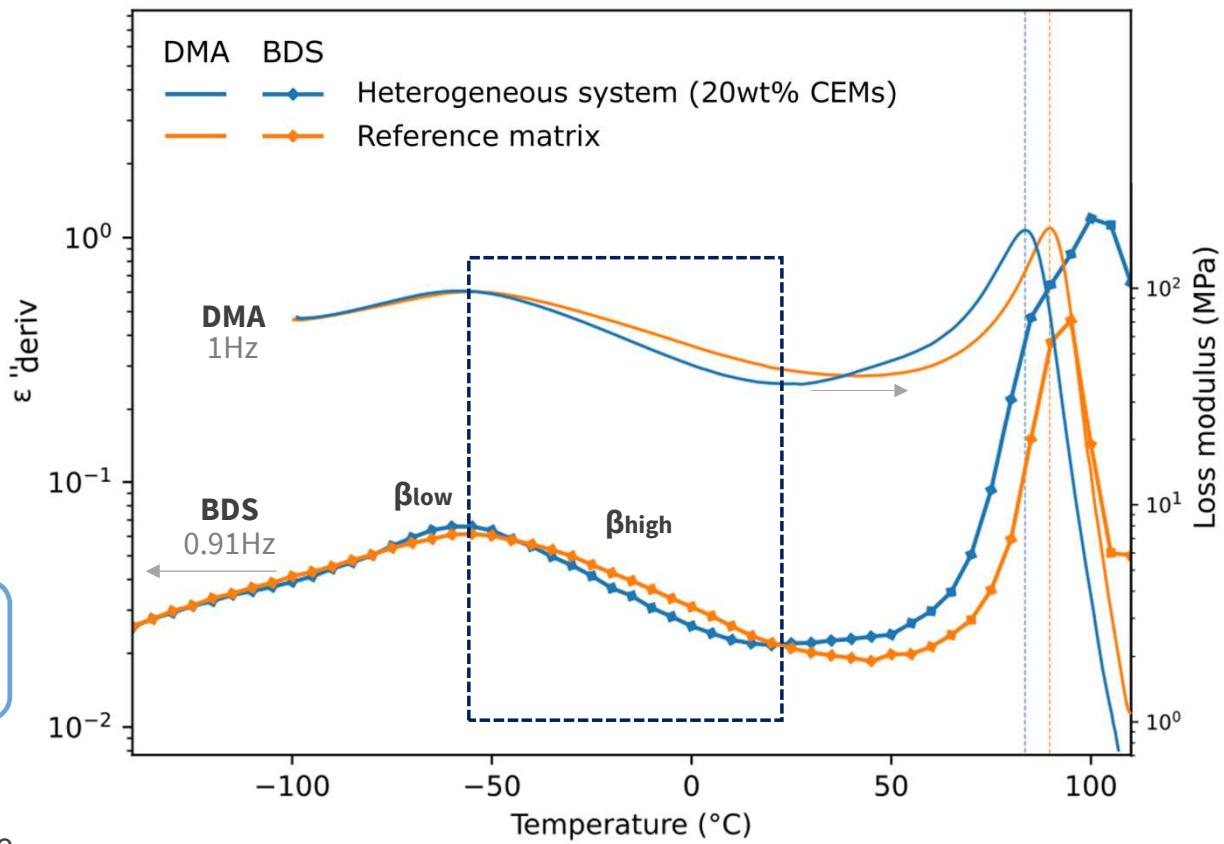
## Isofrequency comparison between DMA and BDS measurements

- $\beta$ -relaxation ascribed to the **localized mobility of hydroxyether units** and/or **phenyl ring flips**
- $\beta_{high}$  was associated with **phenyl ring flips** in the **hardener parts** of the network<sup>1</sup>



- $\beta_{high}$  suppression is **commonly observed** in systems containing structure modification leading to a **lower network crosslink density**<sup>2</sup>

Would CEMs' addition to the network result in lower average cross-linking density of the material?



1. Ramsdale-Capper, R., & Foreman, J. P. (2018). Internal antiplasticisation in highly crosslinked amine cured multifunctional epoxy resins. Polymer, 146, 321-330.

2. Won, Y. G., Galy, J., Gérard, J. F., Pascault, J. P., Bellenger, V., & Verdu, J. (1990). Internal antiplasticization in copolymer and terpolymer networks based on diepoxides, diamines and monoamines. Polymer, 31(9), 1787-1792.

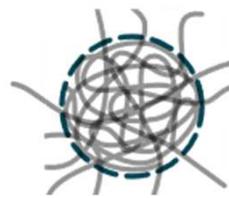
# **Conclusion and perspectives**

## Conclusions

- The obtained results show an impact of the presence of CEMs on both the  $\alpha$  and  $\beta$  modes of the epoxy matrix;
- The single  $T_g$  suggests **homogeneous-like behavior** of the heterogeneous system;
- Meanwhile, **interfacial polarization** observed in BDS measurement suggests **phase heterogeneity** in the heterogeneous system, aligned with SEM observation;
- **$\beta$  high mode suppression** in the heterogeneous system seems to be **due to CEMs'** presence in the matrix. The mechanism generating this **mobility restriction** needs to be elucidated.

## Perspectives

- Finalize **the interpretation of the relaxation modes** in the reference and heterogeneous networks;
- Replicate the study using **different hardeners in the matrix** (e.g. D400 and D2000) to change crosslink density and thus the **contrast between matrix and CEMs**;
- Vary **the fraction of dispersed CEMs** to study the influence of the **heterogeneity degree** on molecular mobility;
- **Network topology** visualization using **AFM**;
- Study the **networks' fracture behavior** (e.g. determine fracture toughness or solvent crack behavior);



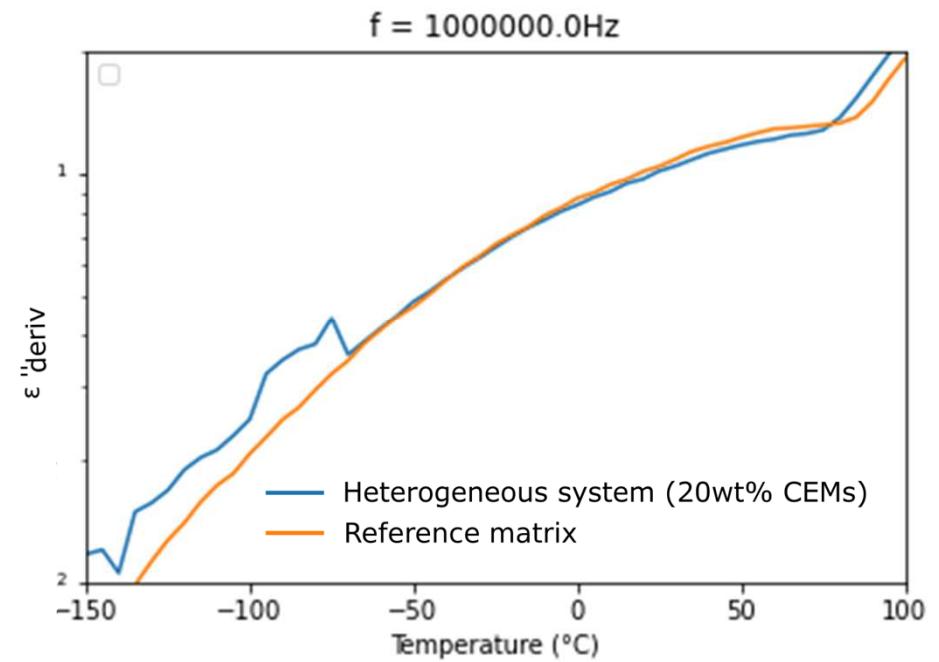
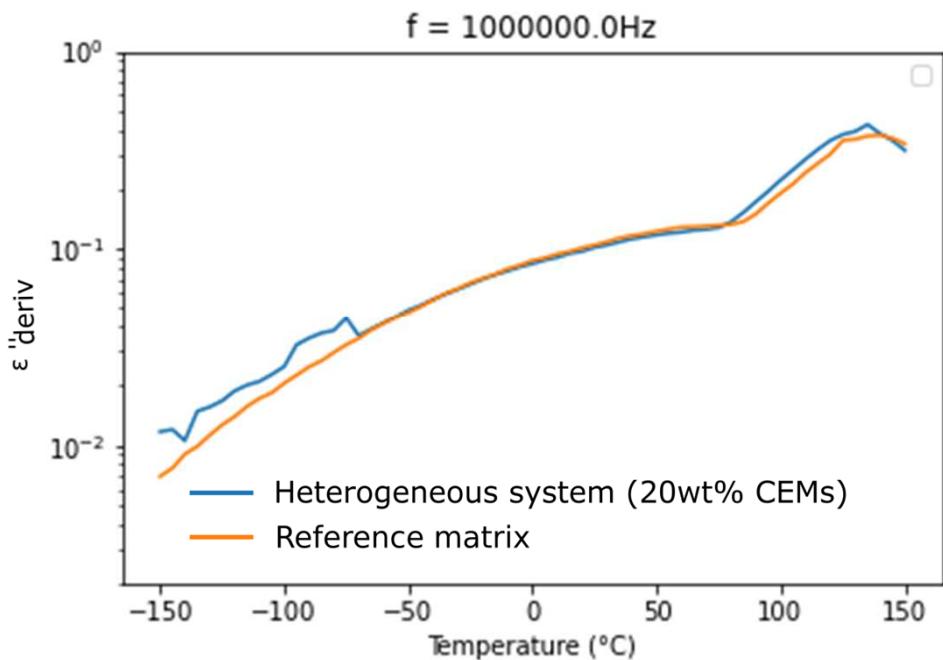
**Thank you for your  
attention**

# **Annexes**

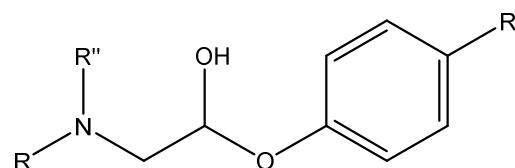
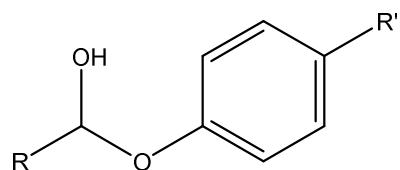
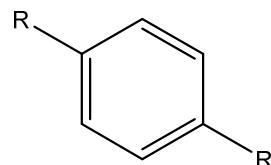
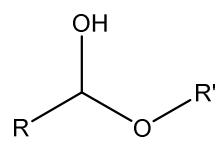
## Other perspectives

- Characterize the films by **IR/NIR, solvent swelling, and pycnometry** to gain more information about the crosslink density;
- Follow the **network build-up** of reference and heterogeneous system by **IR and DSC** to gain information on CEMs' influence on **curing kinetics**
- **Solid NMR** to gain more information about chemical environment of the entities

## Beta relaxation mode shift with increasing frequency



- **hydroxyether** units and/or **phenyl ring** flips

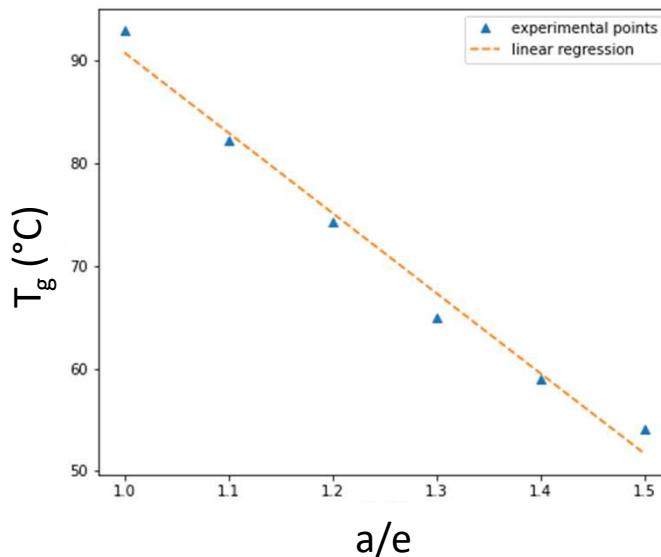


- ◎ **Eps deriv definition**

$$\varepsilon''_{\text{der}} = -\frac{\pi}{2} \frac{\partial \varepsilon'(\omega)}{\partial \ln \omega} \approx \varepsilon''_{\text{rel}}$$

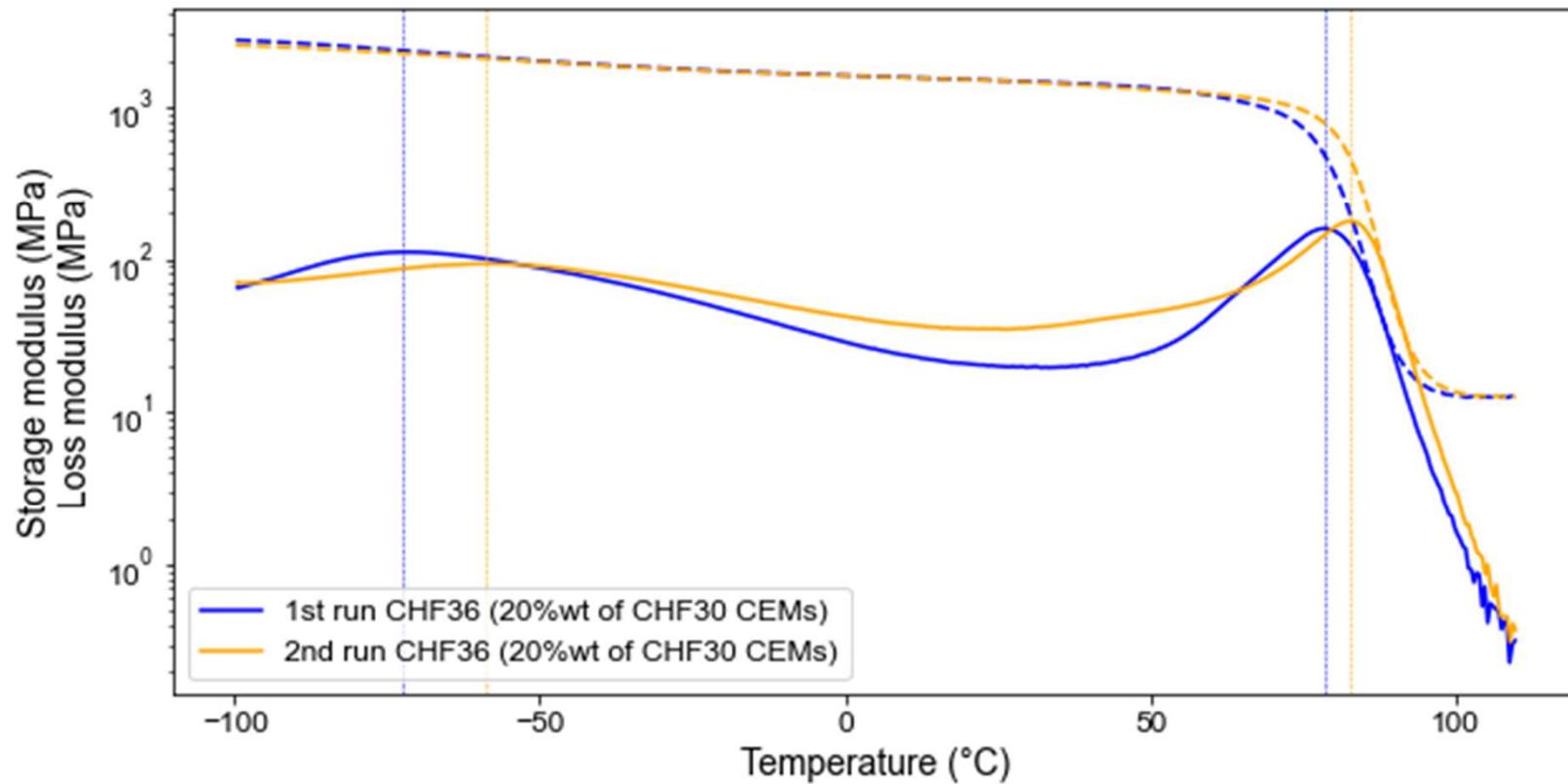
- ◎ “conduction-free” imaginary part,  $\varepsilon''_{\text{der}}$  (that is not entirely true because **ionic conductivity** and **electrode polarization** also contribute to  $\varepsilon'$ ,

## CEMs' final aminohydrogen-to-epoxy ratio estimate

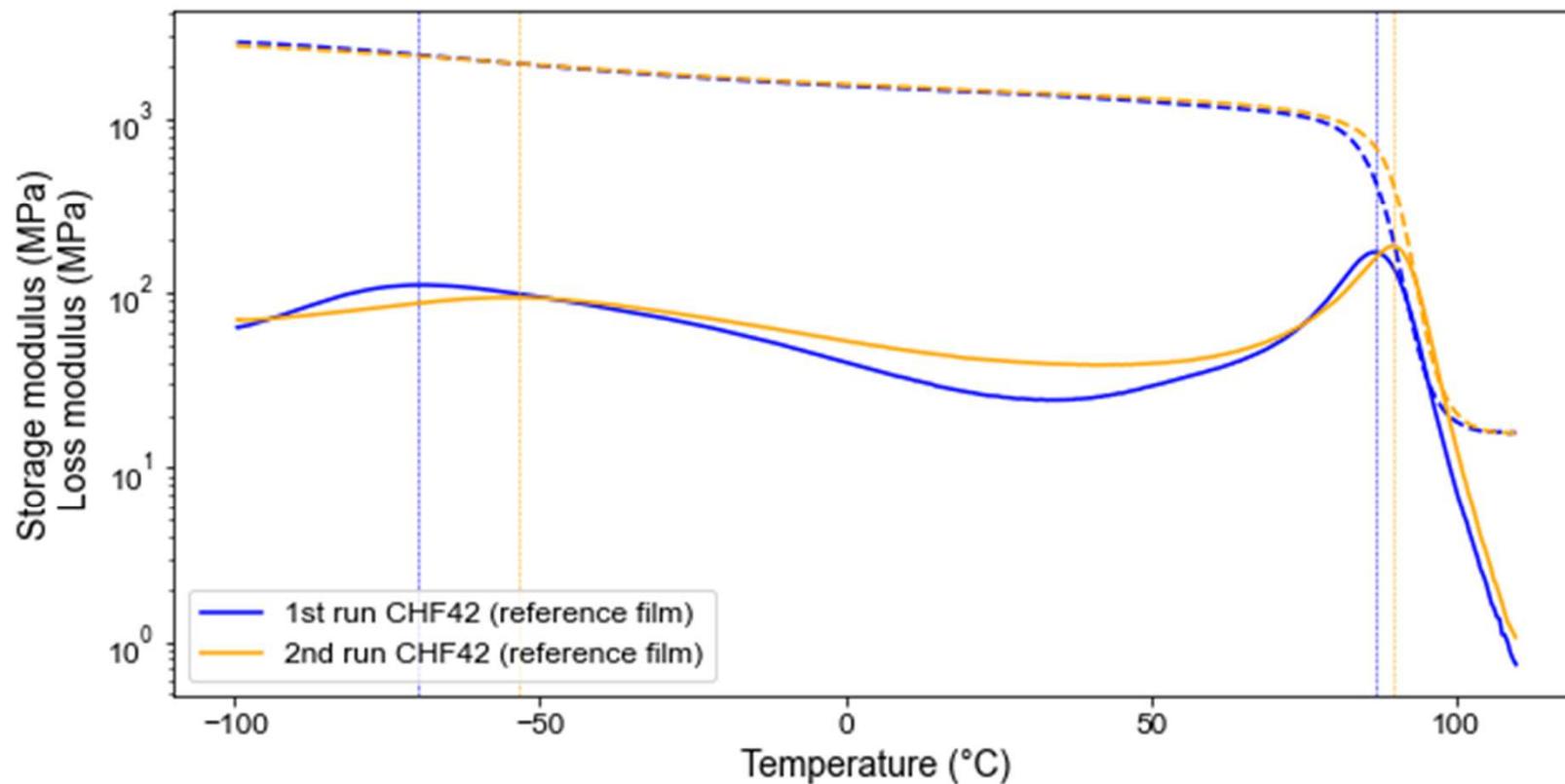


- ◎ Equivalent stoichiometric ratio of microgels calculated from  $T_g$

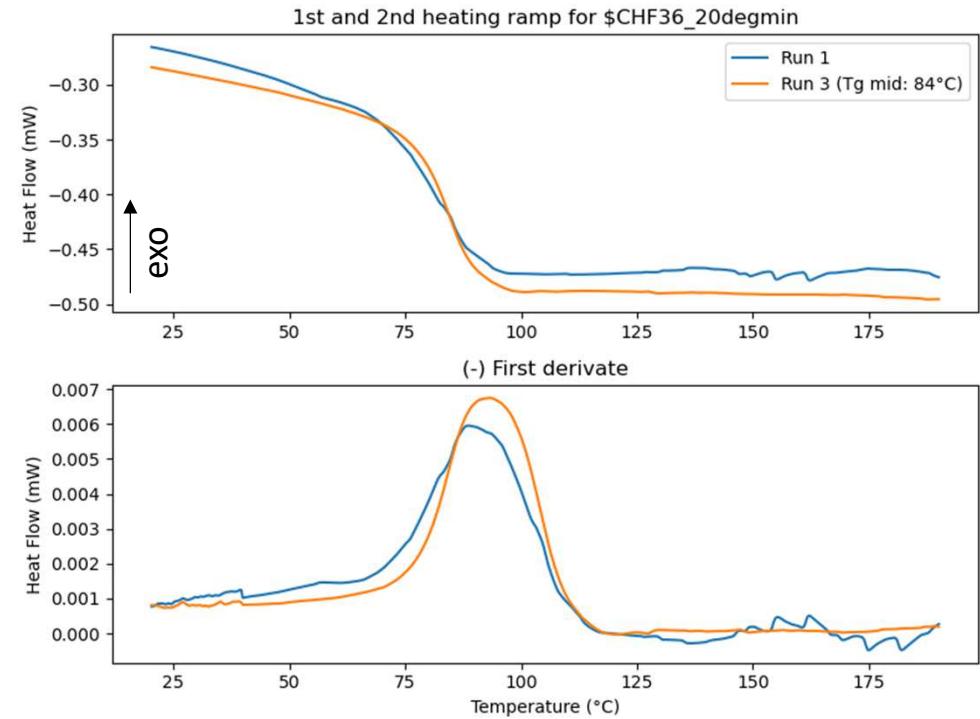
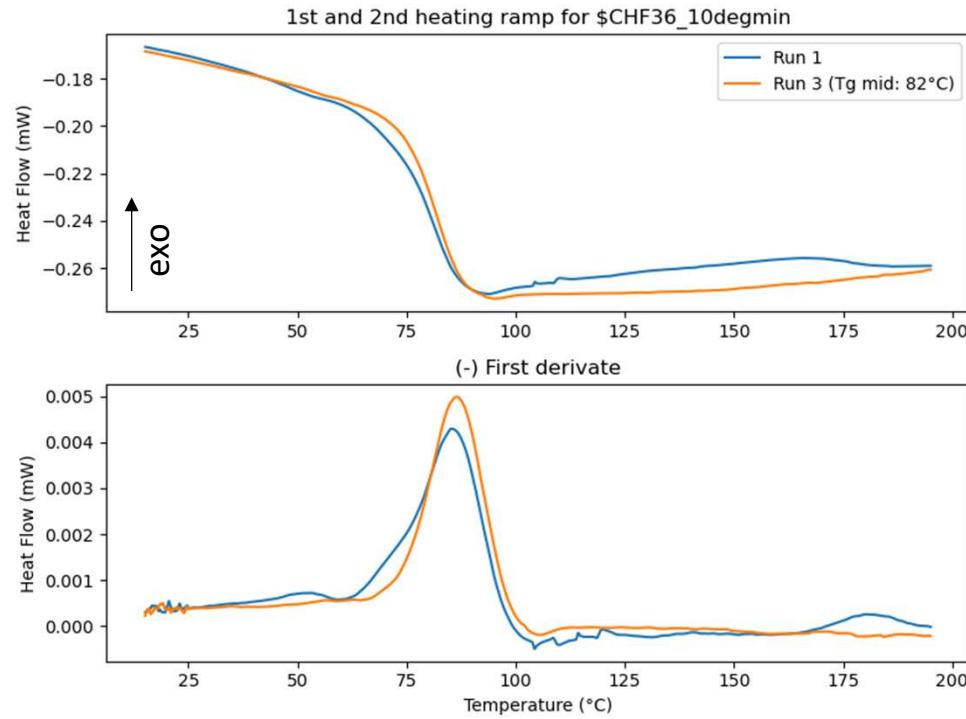
## DMA first and second run: Heterogeneous system film



## DMA first and second run: Reference film

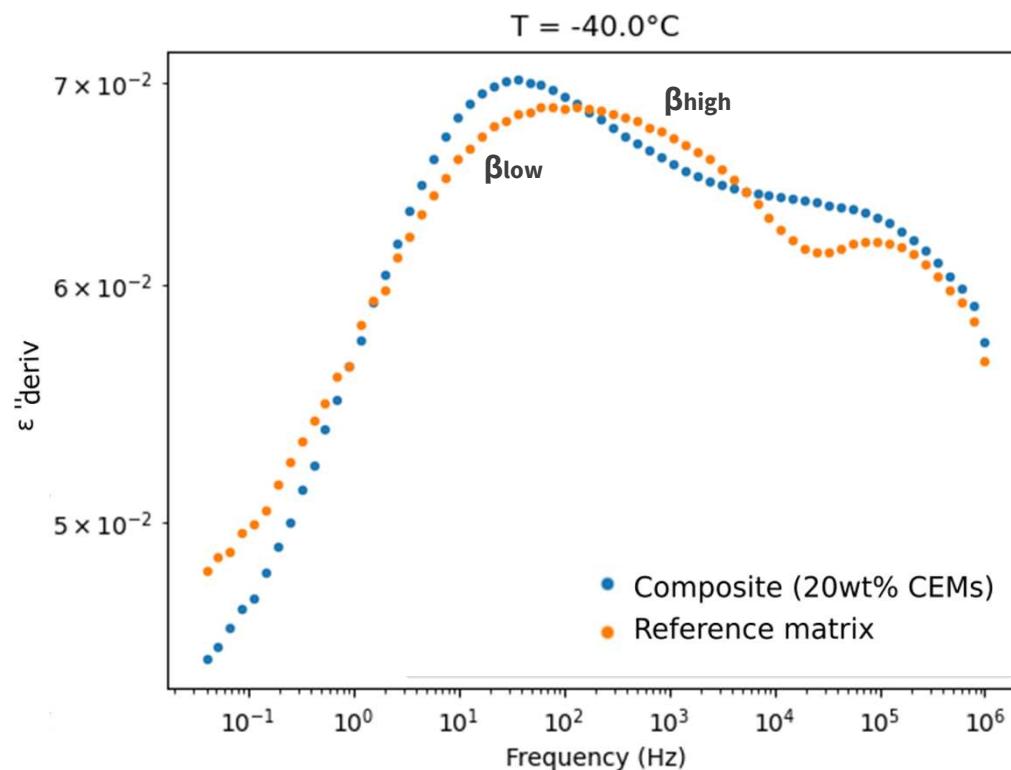


## DSC heterogeneous system (20wt% CEMs): 10 and 20 K/min

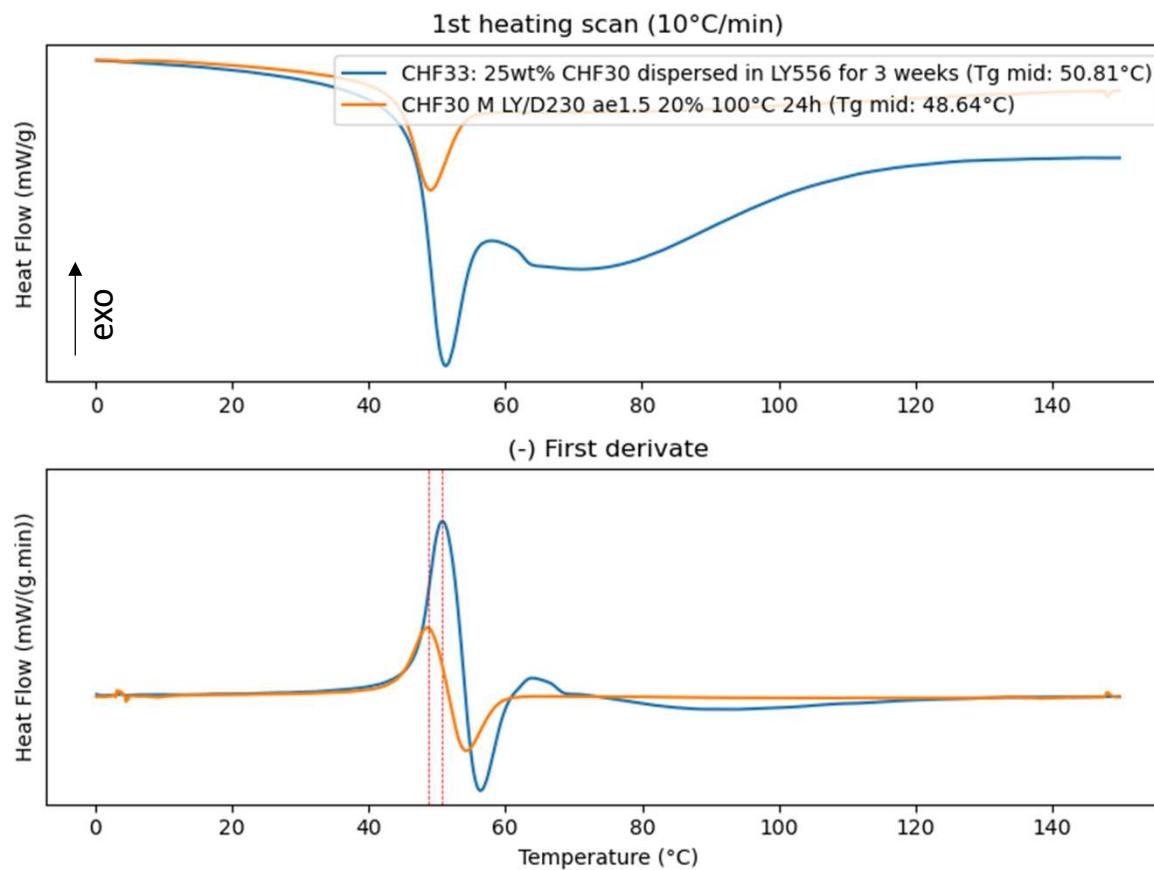


- Single  $T_g$  observed despite of the heating rate.

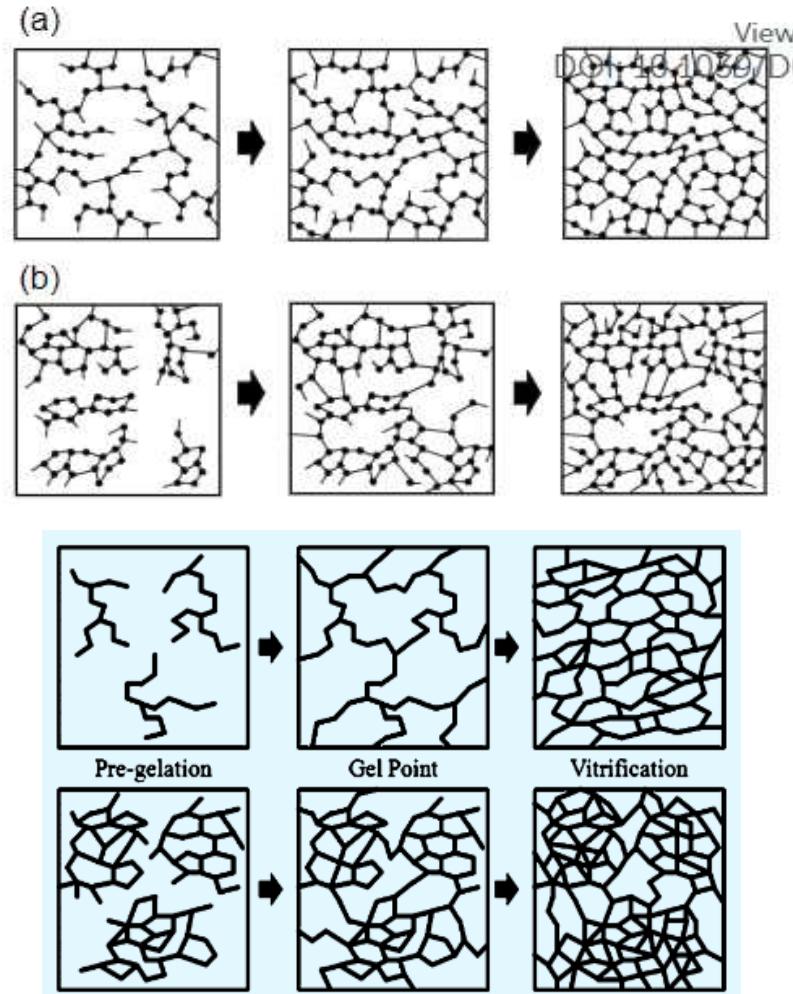
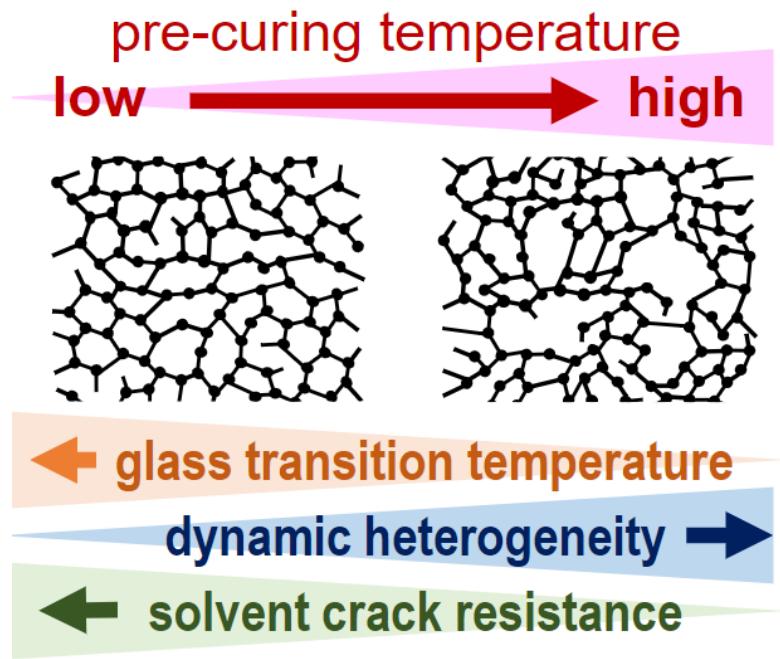
## Isotherm comparison between DMA and BDS measurements



## DMA characterization of reference and heterogeneous system systems



- CEMs T<sub>g</sub> remains the same when embedded in DGEBA for 3 weeks, indicating no significant diffusion of the resin into CEMs core at masterbatch storage conditions



1. Aoki, M., Shundo, A., Yamamoto, S., & Tanaka, K. (2020). Effect of a heterogeneous network on glass transition dynamics and solvent crack behavior of epoxy resins. *Soft Matter*, 16(32), 7470-7478.
2. Sahagun, C. M., & Morgan, S. E. (2012). Thermal control of nanostructure and molecular network development in epoxy-amine thermosets. *ACS applied materials & interfaces*, 4(2), 564-572.